

# DATING METHODS

How many times have you ( heard something like, 'This animal lived 50,000 years ago', or 'This person died 20,000 years ago'? Have you ever wondered how the scientists knew the age of the bone? After all, how old an object is. the scientists haven't been they?

different ways to figure out record. Since it is the written things—it's called carbon.

to tell us the truth about the past. Carefully studying the Biblical record, we find that the universe has an age of around 6,000 years, and that a world-changing, global Flood occurred about 4,300 vears ago.

Those who don't accept the Biblical account of history look for other ways to dis-Of course, the best method cover the age of things. One around that long, have is to check the account of a of these methods is based reliable eyewitness, if one is on a substance found in our There are a variety of available. The Bible is such a bodies, plants and all living

affect how much 14C an animal (or a person or plant) has in it when it dies. This changes how long ago the animal appears to have died. For instance, plants don't Word of God, we can trust it

take in as much 14C as scientists expect. So, after they die, there is less <sup>14</sup>C in the plants to change back to nitrogen. This makes the plant appear to have died many more years ago than it actually did (for example, the plant might appear to be, say, 3,000 years old, rather than 2,000).

There are many factors that

Also, the amounts of <sup>14</sup>C and <sup>12</sup>C in the atmosphere haven't been constant throughout history (for instance, Noah's Flood lowered the total amount of available carbon by burying lots of animals and plants). So

The problem is ...

something that lived (and died) when the proportion of <sup>14</sup>C was

less than normal would appear to have died more years ago than it actually did (for example, it might give an age of 3,000 years before the present, rather than its true age of 2,000 years).

Even many archaeologists don't think 'carbon dating' is completely accurate all the

When these (and other) problems are then taken into account, a scientist can interpret the result of the carbon dating within a Biblical timeframe, but even so, these results can not be used to prove the age of once-living



# HOW CARBON DATING WORKS

There are two basic forms of carbon: one that occurs naturally, called carbon-12 (12C), and one that forms from processes acting on nitrogen in the atmosphere, called carbon-14 (14C). Both of these combine with oxygen to form carbon dioxide (CO<sub>2</sub>), which we breathe out and plants take in. When a cow eats grass, its

of the plant eaters

body absorbs the carbon (both <sup>12</sup>C and <sup>14</sup>C) in the plant.

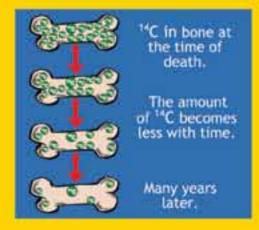
When the cow dies, it stops taking in carbon (for obvious reasons). The amount of <sup>12</sup>C in the cow's body stays the same after death, but the amount of <sup>14</sup>C changes because it returns to nitrogen.

Intake

As time goes on, the amount of <sup>14</sup>C continues to decrease until nothing is left, which is supposedly about 50,000 years later.



the animal (or plant) died. Sounds like a good idea, doesn't it? But there's a problem (see above).



When a paleontologist finds a bone (or a piece of wood), she can measure the amount of <sup>14</sup>C and <sup>12</sup>C it contains. Based on how much <sup>14</sup>C is left, she can supposedly calculate when

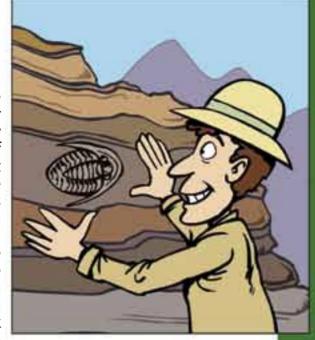


#### ANSWERS FOR KIDS

### How to find a 'date'

Most people think that scientists can actually measure the ages of rocks, using a method called 'radiometric' or 'radioisotope' dating. More often, rocks are 'dated' by the fossils they contain, based on a pre-existing belief in evolution. But even radiometric dating does not actually directly measure the age of something (there is no substance called 'age'). It measures the amounts of certain radioactive substances. This information then has to be interpreted, based on certain beliefs.

In fact, most fossils do not even contain radioactive minerals. So if scientists wanted to measure the age of a fossil using this method, they would look for a nearby layer of igneous rock (e.g. rock that forms from lava from a volcanic eruption)—perhaps in the rock layers above where the fossil was found, or in the layer's below. When they find one, they gather a sample



of the hardened lava and send it off to a laboratory to test it for radioactive elements. The idea behind radiometric dating goes like this (hang on—this gets a bit technical,

but we're sure you'll get it!):

When a volcano erupts, hot, molten rock (called 'lava') from deep inside the Earth is released. This lava is made of various elements. Elements are the 'building-blocks' of the universe (for example, water is made from the elements hydrogen and oxygen).

Some elements (we'll call them 'A') in the lava are radioactive, which means that they change into other elements (we'll call these 'B').

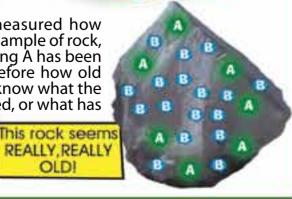


This is often very slow, so that it would take millions (for some elements, billions) of years, starting with a lump of A, for half of it to change into B (the 'half life').



OLDI

However, even if we measured how much of A and B were in a sample of rock, could we figure out how long A has been changing into B, and therefore how old the rock is? No! We don't know what the rock was like when it formed, or what has happened to it since. We This rock seems weren't there.



To test this method, some scientists gathered samples from hardened lava at Mount St Helens, which erupted most recently in the early 1980s. The samples, which came from rocks that formed between 1980 and 1986. were sent to a lab and were 'dated' using the potassiumargon (K-Ar) method. The test results ranged from between 340,000 to 2.8 million years

Other scientists collected samples from cooled lava flows from Mt Ngauruhoe, in New Zealand. These rocks are known to be less than 50 years old, because people observed the volcano erupting in 1949, 1954 and 1975. But the lab results indicated that the rocks were up to 3.5 million vears old!

## But is it a good date?

The scientists who interpret these amounts of A and B to conclude that millions of years have passed must first assume three main things about the rock:

- 1. How much A and B was in the rock when it hardened.
- 2. A has decayed into B at the same rate over
- 3. The amount of A or B in the rock has not increased or decreased in any other way.

But because we haven't been able to study all the rocks everywhere all the time, it's impossible for us to know:

- 1. If B was in the rock before it hardened.
- 2. If A has always decayed at the same rate.
- 3. If water, for instance, has removed some A, or carried some B into the rock from elsewhere.

All of these factors will affect how 'old' the rock appears—in practice, usually making it appear a lot older than it really is.

Known age results: Unknown age: 50 years 350,000



If this method doesn't work on rocks the ages of which we know, how can we trust it to work on rocks of unknown age?

After examining the assumptions behind this 'dating method', and doing scientific experiments to see if this method works on rocks of an already known age, we find that 'radiometric dating' isn't all it's cracked up to be. (Actually, any process used to find ages for things is based on assumptions, and so is not reliable.)

It's important that we allow God's written record of history, the Bible, to guide our thinking about the past—this includes our understanding of the age of the Earth/universe and the age of fossils.

